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Chemistry 3 (Practical) (Alternative A) 12.00pm – 2.00pm (2nd Set) 1.  
A is 0.100 mol dm<sup>-3</sup> solution of an acid. B is a solution KOH  
containing 2.8 g per 500 cm<sup>3</sup>. (a) Put A into the burette and titrate  
it against 25.0 cm<sup>3</sup> portions B using methyl orange as indicator.  
Repeat the titration to obtain consistent titres.

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(Alternative A) 09.30am – 11.30am (1st Set) Chemistry 3 (Practical)  
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solution of an acid. B is a solution KOH containing 2.8 g per 500  
cm<sup>3</sup>. (a) Put A into the burette and titrate it against 25.0 cm<sup>3</sup>  
portions B

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