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Use the following passage and table to answer questions 11–14. A
student studying electron configurations in atoms created Table 1 in
order to summarize the various facts she learned about principal

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energy levels, energy sublevels, and orbitals. 11. Based on Table 1 what is the relationship between the principal energy level number (n)

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59 a student conducted a series of experiments to determine how the solubility of several solid substances varied with temperature she tested.

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